

ME 24-221
Thermodynamics I

Solution to: Quiz 2
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Given: A closed rigid tank
Initial Pressure $P_1 = 1 \text{ Mpa}$
Specific Volume $v_1 = 0.23268 \text{ m}^3/\text{kg}$

Final temperature $T_2 = 165 \text{ }^\circ\text{C}$

Solution:

From Table B.1.2, at 1Mpa, the saturation temperature is $179.91 \text{ }^\circ\text{C}$ and v_g is 0.19444, while the given v_1 is $0.23268 \text{ m}^3/\text{kg}$ ($>v_g$). Hence it is in Superheated state.

From Table B.1.4, at 1 MPa and $v_1 = 0.23268 \text{ m}^3/\text{kg}$, $T = 250 \text{ }^\circ\text{C}$ -----(1)

It is cooled to $165 \text{ }^\circ\text{C}$. At this temperature, the v_f is $0.001108 \text{ m}^3/\text{kg}$ and v_g is $0.27269 \text{ m}^3/\text{kg}$. Since it is a closed rigid tank the specific volume is constant.

$v_f < v_1 < v_g$. Therefore, it is saturated state.

Hence the pressure is the saturation pressure. From Table B.1.2, the saturation pressure at $165 \text{ }^\circ\text{C}$ is, $P_2 = 700.5 \text{ kPa}$. -----(2)

Quality, $x = (v_1 - v_f)/v_{fg}$
 $= (0.23268 - 0.0011008)/0.27158$

$x = 0.853$ -----(3)



