

Multiple Choice:

1. A protein that binds two ligands in a non-cooperative manner will:
 - a) show a hyperbolic binding curve.
 - b) show a curved Scatchard Plot
 - c) show a curved Hill Plot.
 - d) show a sigmodial binding curve
2. Once a ligand dissociation constant (K_D) has been determined it is possible to calculate:
 - a) the ligand binding constant (K_a).
 - b) the ΔG for the binding interaction.
 - c) the concentration of ligand required for half-maximal occupancy
 - d) All of the above are correct
3. In both hemoglobin and myoglobin the oxygen is bound to:
 - a) the nitrogen atoms on the heme.
 - b) polar pocket in the protein.
 - c) histidine residues in the protein.
 - d) The iron atom in the heme group.

Scatchard Plot Fill-In:

The Scatchard equation is:

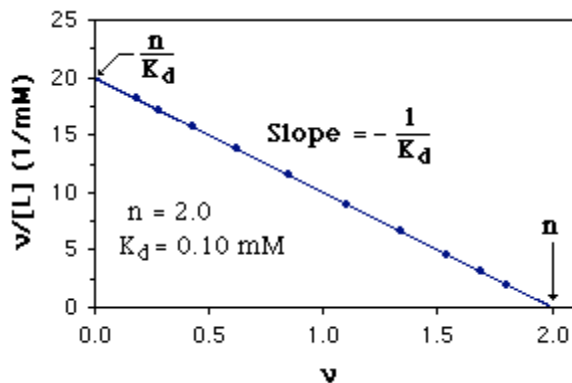
$$\frac{v}{[L]} =$$

$$\frac{L}{K_d} =$$

$$n =$$

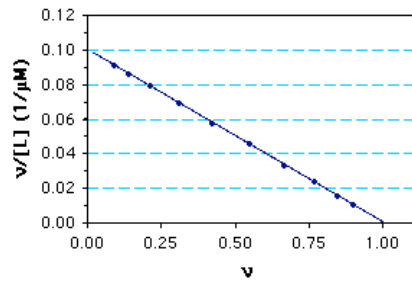
$$K_d =$$

Label the locations of $-1/K_d$, n , and n/K_d on this curve and calculate K_d and



n.

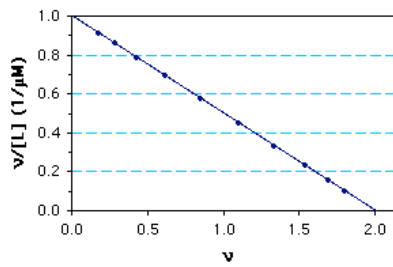
Ligand Binding Curve Problems:



What Type of plot?

N=

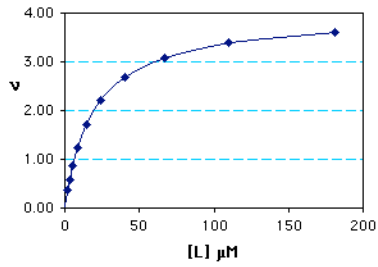
Kd =



Type of Plot:

N=

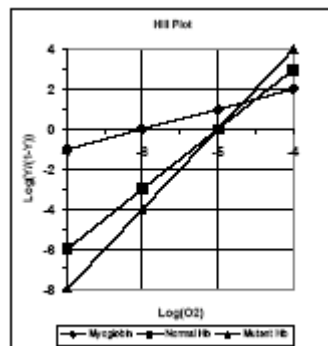
Kd=



Type of Plot?

n=

Kd=



Hill Plot problem:

Determine the Hill coefficient and K_d for the mutant hemoglobin. Please describe your approach