

# 1. Phospholipids

# A. Structure & Nomenclature

**Functions of Biological Membranes** 

Two fatty acids + glycerol + phosphate + a head group form a phospholipid. Various **head groups** are attached to the phosphate, giving a diverse set of lipids

Head group(-X)	Name of Phospholipid	Net Charge	
none	phosphatidic acid (PA)	-1	
choline	Phosphatidylcholine	0	Кон он
(-C-C-N <sup>+</sup> (CH <sub>3</sub> ) <sub>3</sub> )	(PC)	(zwitterion)	
Serine (linkage via sidechain)	Phosphatidylserine (PS)	-1	
ethanolamine	Phosphatidyl-	0	
(-C-C-NH <sub>3</sub> <sup>+</sup> )	ethanolamine (PE)	(zwitterion)	
inositol $(C_6H_6O_5)$	Phosphatidylinositol (PI)	-1	

# Know the structure of either PC or PS as an example.

# 2. Membrane Proteins

# Peripheral Membrane Proteins:

Loosely attached to membranes via electrostatic interactions – released with high salt.

#### Integral Membrane Proteins:

Largely contained within the membrane (solubilization requires disruption of the membrane by detergents). Often span the entire membrane. Stability energetics are similar to water soluble proteins, except that non-polar groups interact with acyl chains in the membrane. The rule here is: "hydrophobic outside hydrophilic **or** hydrophobic inside", thereby matching the location. Asymmetry across the bilayer is required for most functions (both lipid a





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is required for most functions (both lipid and protein). No flip-flop!

Transport (e.g. of protons, metabolites, electrons)

- <u>Passive</u> transport (no energy required, molecules go from high concentration to low)
- ✤ <u>Active</u> transport (energy required, molecules are pushed from low to high concentration)



# 3. Cholesterol enhances membrane fluidity:

Lipid bilayers undergo a *highly cooperative* phase transition with a defined T<sub>m</sub>:

Below  $T_m$  the the acyl chains are tightly packed and the lipids exist as a solid-like *gel*; . Above  $T_m$  the acyl chains are disordered and the lipids are in a liquid-like *liquid crystal phase*. Rapid lateral diffusion of lipids and proteins occurs in the plane of the membranes above but not below  $T_m$ .



- Membrane fluidity is essential for the biological function of membranes. Therefore organisms go to great lengths to maintain a fluid membrane bilayer (eg, bacteria regulate acyl chain length).
- Lipid bilayers can be made more fluid (higher T<sub>m</sub>) *either* by decreasing fatty acyl chain length *and/or* increasing the degree of unsaturation. (Chain length and the presence of cis-double bonds can greatly affect T<sub>m</sub> and fluidity through affecting the extent of van der Waals packing)



 But simply increasing the degree of unsaturation can compromise the integrity of the cell membrane. Animals cells have another way of increasing membrane fluidity:

# Cholesterol:

- Is a natural steroid, you produce about 1 g/day!
- About the same length as a C<sub>16</sub> fatty acid; therefore it reaches across *half* of the bilayer.
- *Essential* component of most mammalian membranes (~20% of cell membrane lipid)
- Destroys the phase transition of pure lipid membranes, thereby keeping the membranes fluid below the phase transition and more rigid above the phase transition. Often referred to as a membrane plasticizer.





# Fluid mosaic model of Biological Membranes

Lipids, membrane proteins, glycolipids, etc... float in a 2-D fluid membrane.

Rapid lateral diffusion of components (~1 µm/sec!).

Different types of lipids – composition varies (type & acyl chain) by organism, tissue type, and temperature.

The membrane completely encloses the cell (or intracellular organelle) and therefore *defines cellular topology* (*i.e.* an inside and an outside).

- \* Transport of molecules and ions occurs across membranes.
- \* Generation of a concentration difference across the membrane can be used to produce energy.
- \* Signal transduction occurs across the membrane.



