

## Lecture 20: Analysis of Enzyme Inhibition & Protein Purification

Reading in Campbell: Chapter 6.7, chapter 5

### 20.1 Summary of Analysis of Inhibition

A competitive inhibitor binds to the substrate-binding pocket of E with a dissociation constant of  $K_I$ . By competing with substrate for binding to the same site in E, a competitive inhibitor effectively increases the  $K_M$  of the enzyme-catalyzed reaction. Because it does *not* bind to the ES complex, a competitive inhibitor does *not* affect  $V_{MAX}$ .  $\alpha$ , the degree to which a competitive inhibitor increases  $K_M$ , is obtained from the ratio of the slopes (slope (+ inhibitor) / slope (- inhibitor)) on a double reciprocal plot. From  $\alpha$ ,  $K_I$  can be calculated if the inhibitor concentration is known.

$$K_I = \frac{[E][I]}{[EI]}$$

$$\frac{1}{v} = \frac{\alpha K_m}{V_{MAX}} \frac{1}{[S]} + \frac{1}{V_{MAX}}$$

$$\alpha = 1 + \frac{[I]}{K_I}$$

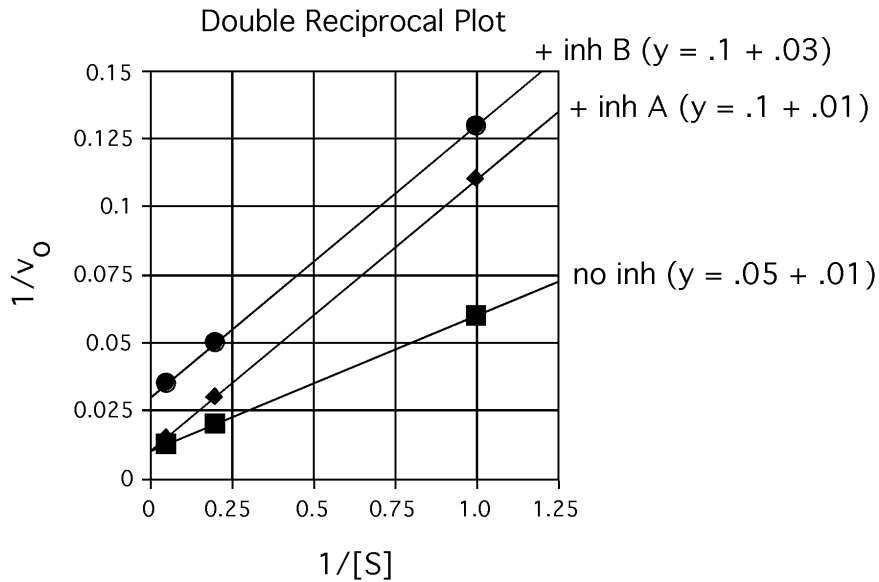
A noncompetitive inhibitor binds to *both* E and ES at a site *distinct* from the substrate binding site. Although a noncompetitive inhibitor usually binds to E and ES at the same site, it does so with differing affinities. It binds E with a dissociation constant of  $K_I$  and ES with a dissociation constant of  $K_I'$ . Binding of a noncompetitive inhibitor to E and ES allosterically inhibits both substrate binding and catalysis, respectively. Thus *both*  $K_M$  and  $V_{MAX}$  are affected.  $\alpha$ , the degree to which a noncompetitive inhibitor increases  $K_M$ , is obtained from the ratio of the slopes (just like for a competitive inhibitor).  $\alpha'$ , the degree to which a noncompetitive inhibitor decreases  $V_{MAX}$ , is obtained from the ratio of the y-ints (y-int (+ inhibitor) / y-int (- inhibitor)).  $K_I$  and  $K_I'$  can be calculated from  $\alpha$  and  $\alpha'$ , respectively, if the inhibitor concentration is known.

$$K_I = \frac{[E][I]}{[EI]} \quad K_I' = \frac{[ES][I]}{[ESI]}$$

$$\frac{1}{v} = \frac{\alpha K_m}{V_{MAX}} \frac{1}{[S]} + \frac{\alpha'}{V_{MAX}}$$

$$\alpha = 1 + \frac{[I]}{K_I} \quad \alpha' = 1 + \frac{[I]}{K_I'}$$

**Sample Analysis:**



**20.2 Protein Purification**

Reading: Campbell Chapter 5

Key Terms:

- **Specific Activity**
- **Fold purification**
- **Purification Scheme**
- **Chromatographic Steps**

**Concept of a Purification Scheme:**

A purification scheme usually begins with a crude mixture of cellular proteins, referred to as a *lysate*. The lysate is treated in a series of physical steps or processes. Each step separates a mixture of proteins into two or more fractions. Fractions that contain the protein or enzyme of interest are retained for the next step of the purification scheme while the other fraction(s) are discarded until the protein is deemed to be pure. The entire sequential process is referred to as a purification scheme.

**The actual separation steps are based on different physical/chemical properties of proteins:**

- Solubility in different salts (ammonium sulfate)
- Size
- Charge
- Hydrophobicity
- Binding to Specific Ligands

**How to monitor purity:** It is essential to have some method to evaluate fold-purification or fold-enrichment, otherwise how do you know whether a specific step in the purification scheme has been successful in increasing the purity of the desired protein?

*Amount of target enzyme:* It is essential to have some method of determining the amount of the desired enzyme at any given step of the purification scheme. Consequently, the most critical step in any purification scheme is to develop a suitable assay for the enzyme that is being purified.

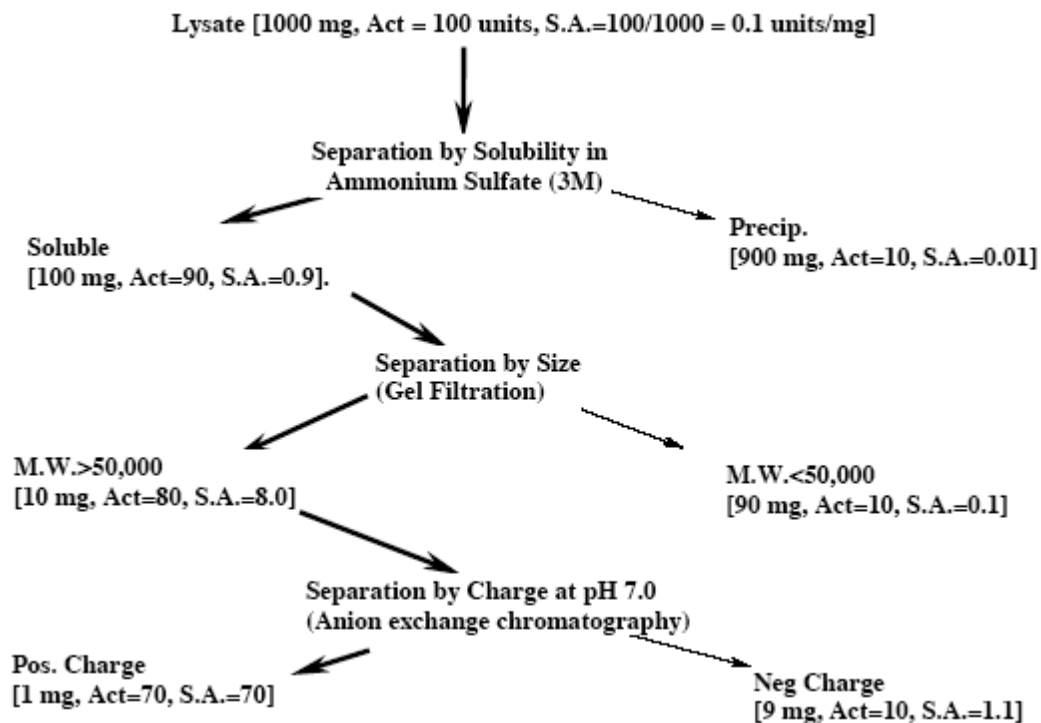
*Measuring purity during purification:* The purity of the enzyme during the purification scheme is generally monitored by measuring the specific activity. Specific activity is defined as:

**Specific Activity:** The units of activity (eg,  $V_{MAX}$  in the case of an enzyme catalyzed reaction) for the desired enzyme divided by the total amount of *all* protein species in the sample. Typical units would be  $\mu\text{mol}/\text{sec}/(\text{mg total protein})$ , where the  $\mu\text{mol}/\text{sec}$  refers to the amount of product produced ( $\mu\text{mol}$ )/unit time. As the purification scheme progresses, the specific activity should *increase*, reaching a maximum for the pure enzyme.

**Example Purification Scheme:**

The following scheme is an example of purification of a large (i.e. > 50,000 Da) protein that is soluble in 3M ammonium sulfate and has a positive charge at pH=7.0.

- Initially, we begin with 1 gram of total protein which contains 100 units of activity, giving an initial specific activity of 0.1 units/mg.
- The final yield is 1 mg of purified material with a specific activity of 70 units/mg. Since the total activity of the final material was 70 units, the yield of this purification scheme is 70% (70 units/100 units).



**Example purification scheme:** Garden beets were placed into a Waring blender to produce an initial lysate for the purification of rubisco, an important plant enzyme. The following data were obtained for each step of the purification:

Sample	Units of Activity $\mu\text{mol product/sec}$	Total Protein (mg)	Specific Activity $\mu\text{mol product/sec/mg}$
1. Crude lysate	50,000 $\mu\text{mol/sec}$	5,000 mg	10
2. After separation by solubility in ammonium sulfate.	45,000 $\mu\text{mol/sec}$	900 mg	50
3. After separation by size.	40,000 $\mu\text{mol/sec}$	800 mg	50
4. After separation by charge.	30,000 $\mu\text{mol/sec}$	10 mg	3000

- i) Calculate the net % yield:
- ii) Calculate the increase in purity:
- iii) Which step in the above purification scheme was a complete waste of time? How do you know?
- iv) Is the protein pure after the last step?

#### Evaluating Final Purity:

After the protein is pure, its purity can be monitored by:

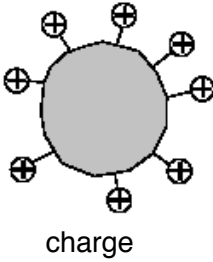
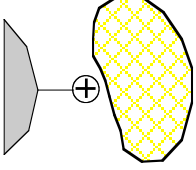
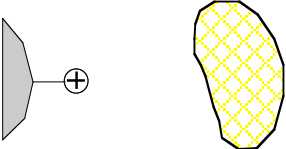
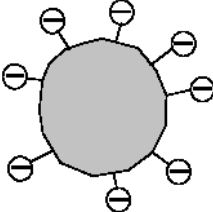
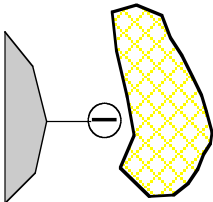
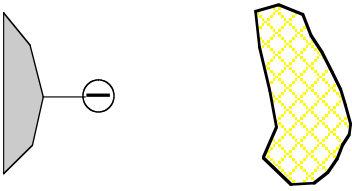
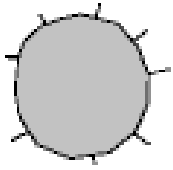
- a) SDS-page gel electrophoresis.
- b) Mass spectrometry.
- c) Amino terminal sequencing.

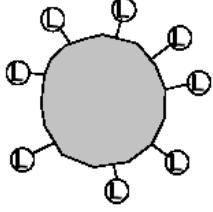
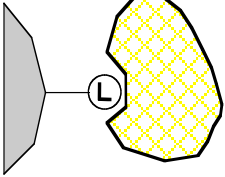
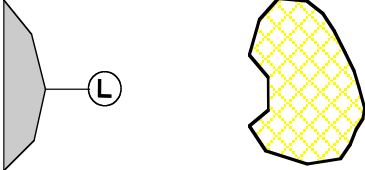
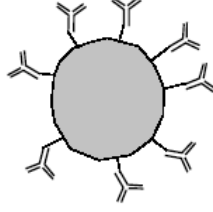
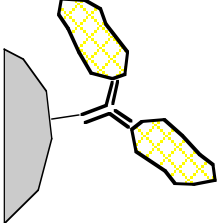
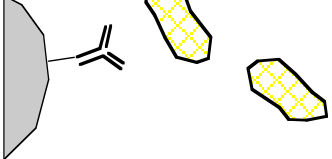

#### Separation using Column Chromatography:

In most cases chromatography is performed in long glass tubes filled with a matrix or resin (particle size similar to a fine sand) that is completely immersed in a buffered salt solution. The mixture of proteins is added to the top of this column and buffer is allowed to flow through the column. As the buffer flows through the column the mixture of proteins is drawn down through the column and interacts with the matrix or resin. The actual mode of separation depends on the nature of the resin. Usually several different chromatographic steps are performed with different resins during a purification scheme. The actual order of separation methods will depend on the protein being purified.

#### Separation by Binding:

- Anion exchange
- Cation exchange
- Hydrophobicity
- Affinity chromatography

Type	Type of Resin	Principal of Separation	How to Elute the Protein
<p><b>Anion Exchange</b></p> <p><i>Separation by charge.</i></p>	<ul style="list-style-type: none"> <li>Beads with a <b>positive</b> charge</li> </ul> 	<p>Protein sticks to resin because of:</p> <ul style="list-style-type: none"> <li>Overall <b>negative</b> charge (<i>anions</i>)</li> <li>Proteins have patches of negative charge</li> </ul> 	<ul style="list-style-type: none"> <li>Increase salt concentration to weaken electrostatic interaction.</li> <li>Change of pH to <math>\text{pH} &lt; \text{pI}</math> (protein becomes positively charged)</li> </ul> 
<p><b>Cation Exchange</b></p> <p><i>Separation by charge.</i></p>	<ul style="list-style-type: none"> <li>Beads with a <b>negative</b> charge</li> </ul> 	<p>Protein stick to resin because of:</p> <ul style="list-style-type: none"> <li>Overall <b>positive</b> charge (<i>cations</i>)</li> <li>Proteins have patches of positive charge</li> </ul> 	<ul style="list-style-type: none"> <li>Increase salt concentration to weaken electrostatic interactions.</li> <li>Change of pH to <math>\text{pH} &gt; \text{pI}</math> (protein becomes negatively charged)</li> </ul> 
<p><b>Hydrophobic Chromatography</b></p> <p><i>Separation by hydrophobicity</i></p>	<ul style="list-style-type: none"> <li>Beads with a <b>hydrophobic</b> group</li> </ul> 	<p>Protein sticks to resin in high salt because of:</p> <ul style="list-style-type: none"> <li><b>Hydrophobic</b> patches on surface (salts like ammonium sulfate remove water ordered around hydrophobic patches on protein surface and expose them for interaction with the resin)</li> </ul>	<ul style="list-style-type: none"> <li>Decrease salt concentration to weaken hydrophobic interactions (hydrophobic patches are once again surrounded by water)</li> </ul>

<p><b>Affinity Chromatography</b></p> <p><i>Separation by affinity.</i></p>	<ul style="list-style-type: none"> <li>• Beads with a ligand:</li> </ul> 	<p>Protein stick to resin because of:</p> <ul style="list-style-type: none"> <li>• Binding site for ligand</li> </ul> 	<ul style="list-style-type: none"> <li>• Excess ligand</li> <li>• Change in pH, salt, solvent to weaken protein-ligand interaction.</li> </ul> 
	<ul style="list-style-type: none"> <li>• or Antibody</li> </ul> 	<p>Protein stick to resin because of:</p> <ul style="list-style-type: none"> <li>• Binding to antibody</li> </ul> 	<ul style="list-style-type: none"> <li>• Changes in solution conditions (pH, salt, solvent) to weaken protein-antibody interaction.</li> </ul> 
<p><b>Gel Filtration:</b></p> <p><i>Separation by size</i></p>		<ul style="list-style-type: none"> <li>• Proteins don't "stick".</li> <li>• Small proteins enter the interior of the beads, and therefore take longer to wash off of the column.</li> </ul>	<ul style="list-style-type: none"> <li>• Simply washing the column with buffer will eventually wash the proteins out of the column.</li> <li>• Smaller proteins elute last.</li> </ul>