

## Objective

- Quantify the **ammonia emissions** from one of the major potential carbon capturing processes, **amine scrubbing**.
- Evaluate the implications for **air quality**, focusing on the impact on PM<sub>2.5</sub>.

## 1. Background

- Carbon Capture and Storage (CCS) is a potential strategy for reducing CO<sub>2</sub> emissions at coal power plants.
- Amine scrubbing is one of the most proven CCS technologies currently available [1].
- The major potential environmental concerns of amine scrubbing are spent solvent, amine and NH<sub>3</sub> emissions [2].
- An aggressive deployment of amine scrubbing may increase NH<sub>3</sub>, a PM<sub>2.5</sub> precursor, in the atmosphere.

## 2. NH<sub>3</sub> emissions and CCS in 2050

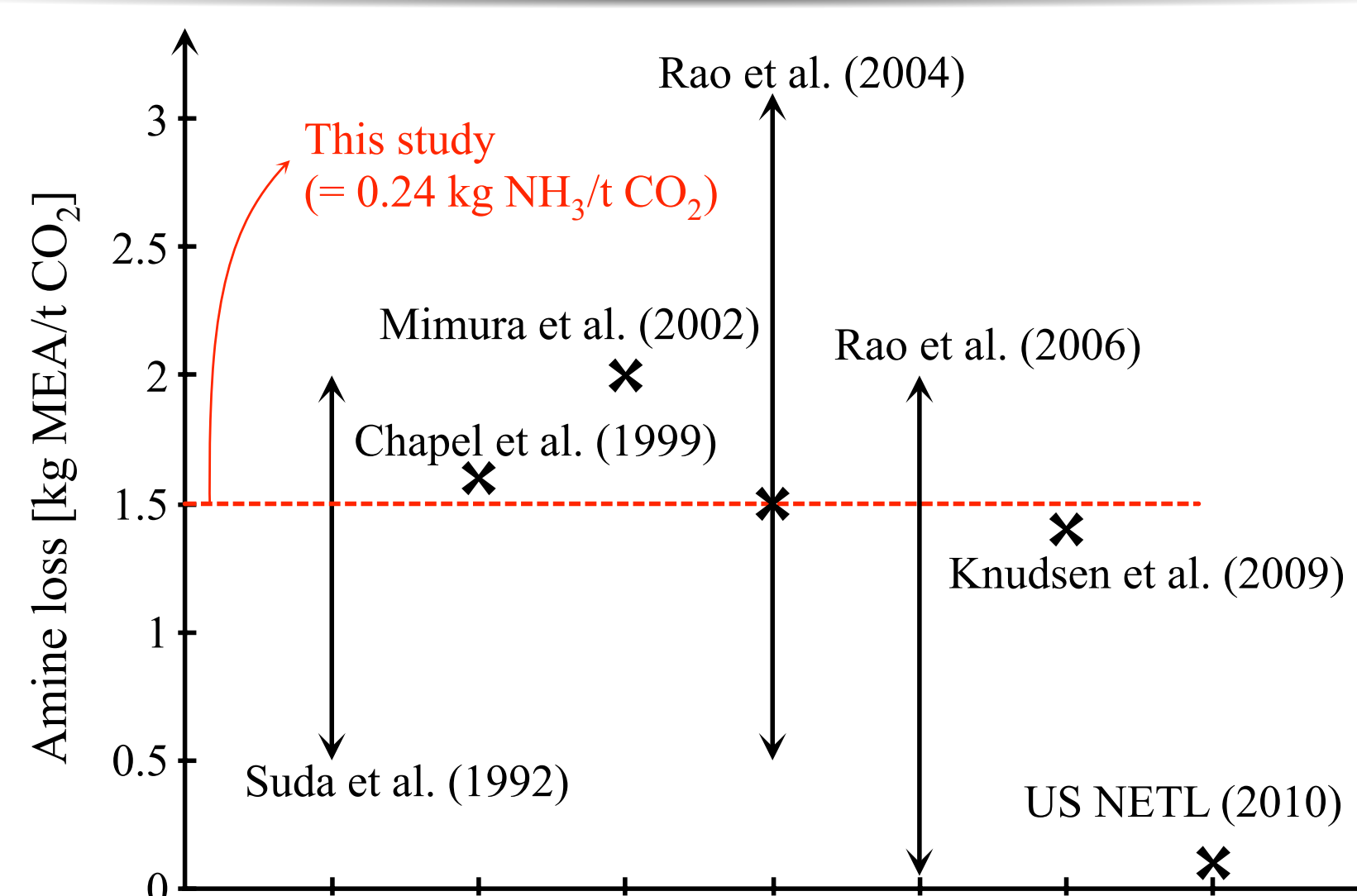


Figure 1: Amine loss

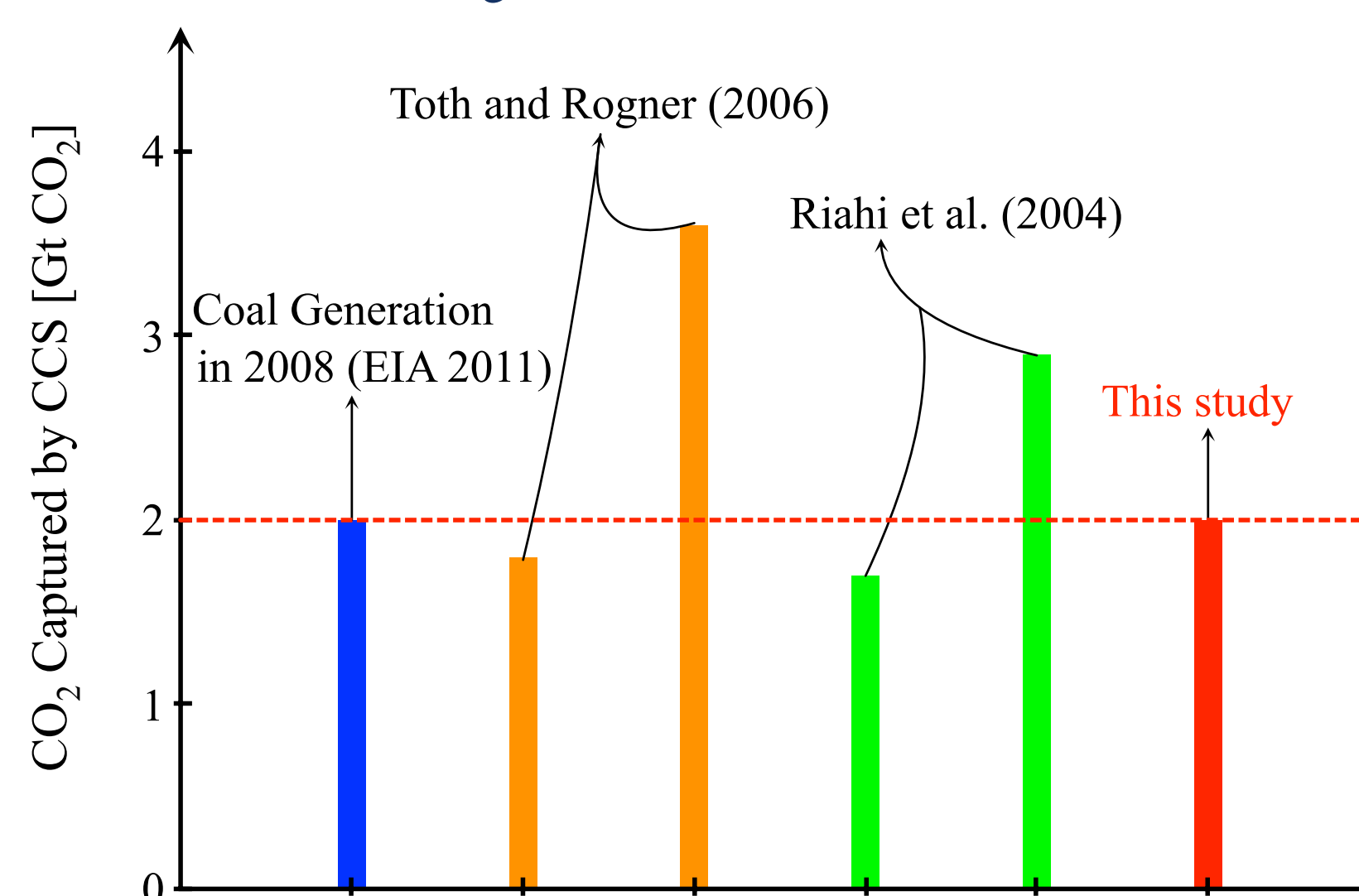


Figure 2: US CCS deployment potential in 2050

- US NH<sub>3</sub> emissions from CCS in 2050  
= (NH<sub>3</sub> Emissions Factor) × (CO<sub>2</sub> captured by CCS)  
= 0.43 Tg N/year in the Eastern US

## 3. Scenarios

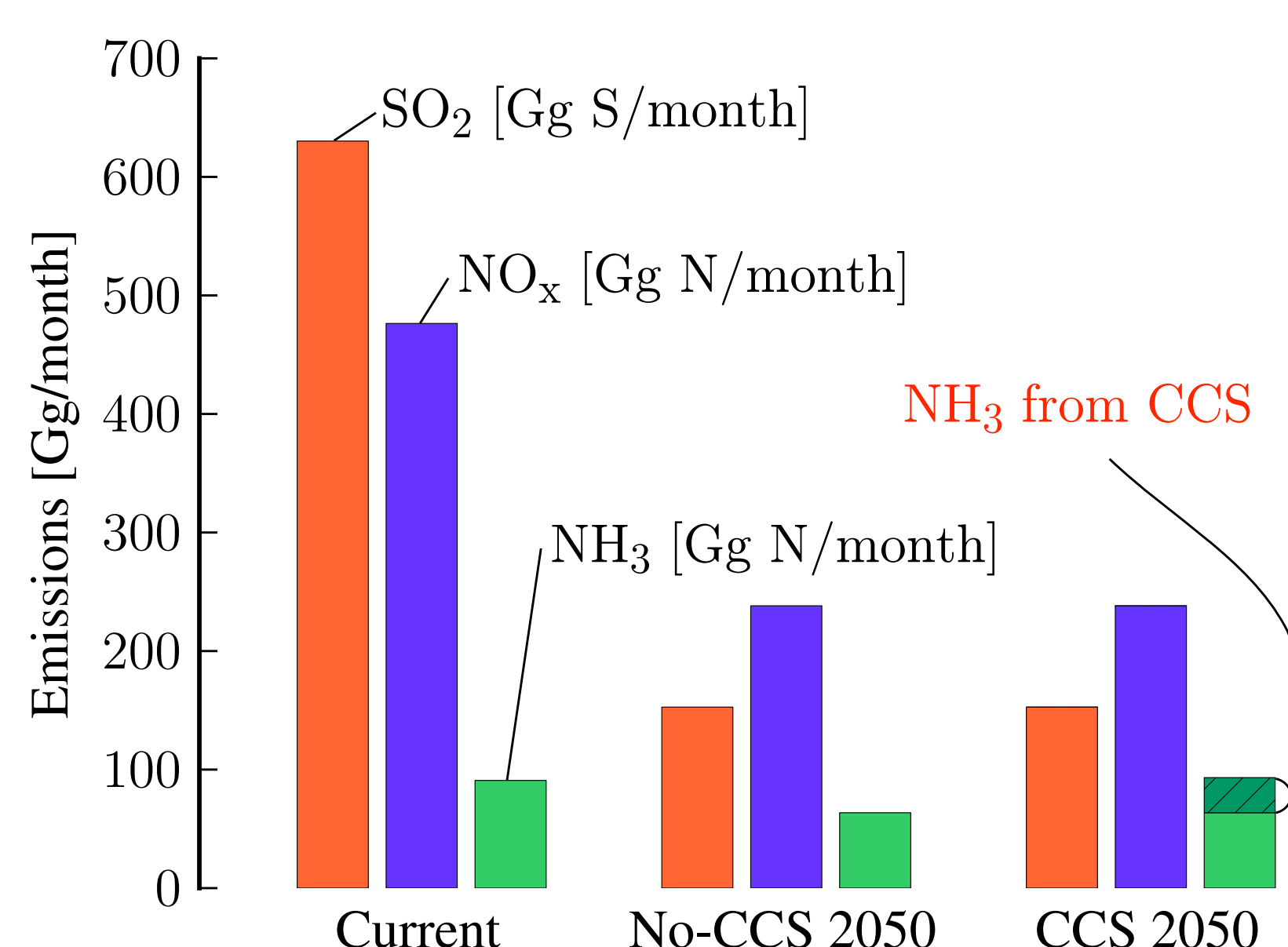


Figure 3: January emissions assumptions of three scenarios

- Current:** represents current emissions as of 2001-2002.
- No-CCS 2050:** Current and future air quality regulations reduce 80% of SO<sub>2</sub>, 50% of NO<sub>x</sub>, and 30% of NH<sub>3</sub>.
- CCS 2050:** In addition, coal power plants with amine scrubbing CCS capture 2.0 Gt CO<sub>2</sub>/year.

## 4. PM<sub>2.5</sub> and Ammonia

- PM<sub>2.5</sub>, particulate matter having a diameter of 2.5 μm or less, is known to pose the greatest human health risks.
- NH<sub>3</sub> reacts with SO<sub>2</sub> and NO<sub>x</sub> **non-linearly** to form PM<sub>2.5</sub>.
- PM nitrate (NH<sub>4</sub>NO<sub>3</sub>) formation may significantly increase PM<sub>2.5</sub> concentrations in winter in the US [3].

Table 1: PM<sub>2.5</sub> nitrate formation governing conditions.

NH <sub>3</sub> availability	PM <sub>2.5</sub> nitrate form?	Limited by
Limited	No	-
Moderate	Yes	NH <sub>3</sub>
Excess	Yes	HNO <sub>3</sub>

## 5. Results of Air Quality Simulations

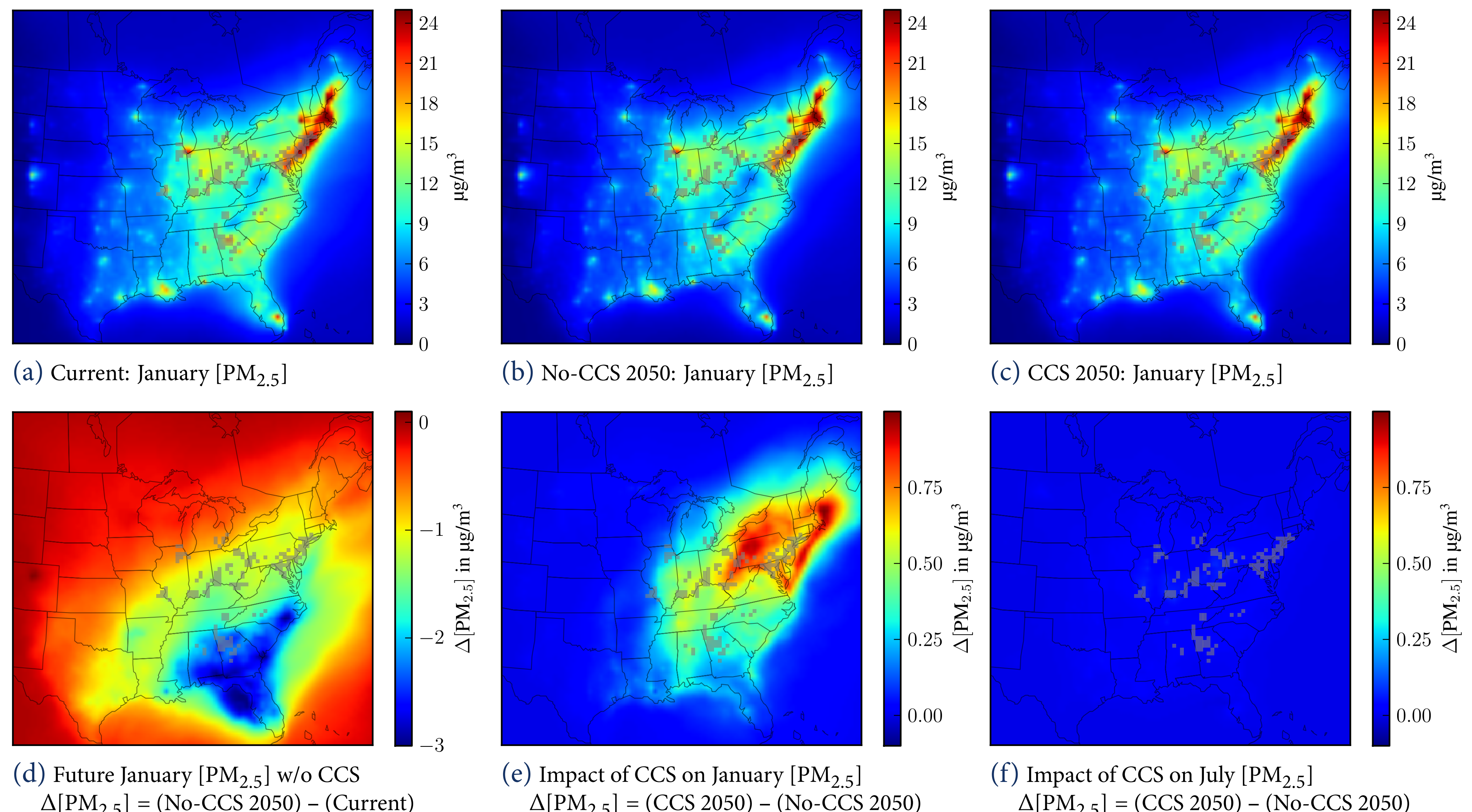


Figure 4: Air quality simulation results from PMCAMx, a 3D chemical transport model. The average PM<sub>2.5</sub> increase in nonattainment areas (Gray dots) is 0.53 μg/m<sup>3</sup> in January and 0.04 μg/m<sup>3</sup> in July.

## 6. Sensitivity Analyses

Table 2: Emissions assumption of two sensitivity scenarios, which capture the uncertainty of future emissions.

Scenarios for 2050	SO <sub>2</sub>	NO <sub>x</sub>	NH <sub>3</sub>
No-CCS 2050	80%	50%	30%
High-sensitivity	90%	20%	50%
Low-sensitivity	30%	70%	0%

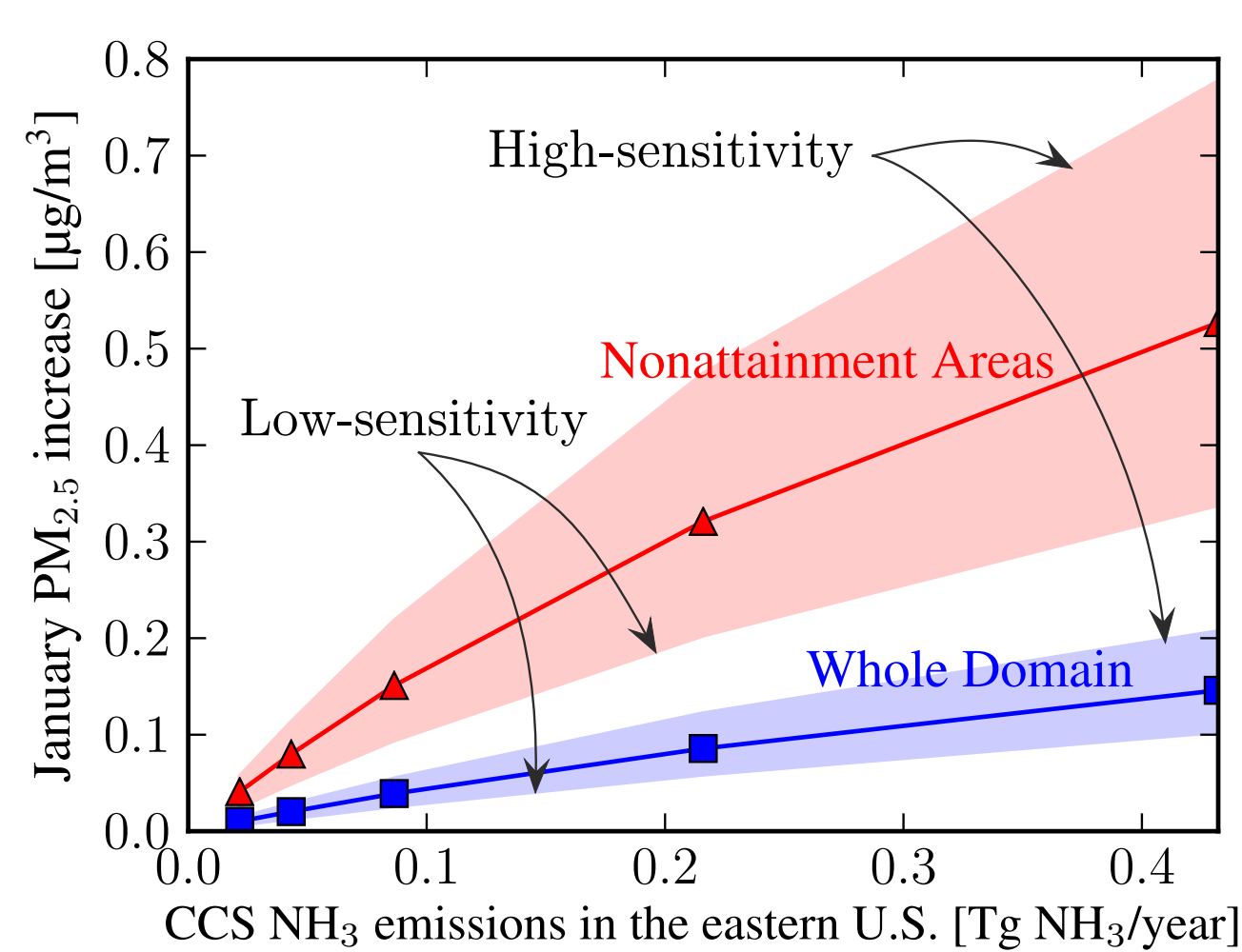


Figure 5: Sensitivity of January PM<sub>2.5</sub> increase to two major uncertainties, NH<sub>3</sub> emissions and future air quality.

## 8. Conclusions

- January** PM<sub>2.5</sub> may increase by 0.5 μg/m<sup>3</sup> on average and up to 0.9 μg/m<sup>3</sup> in PM<sub>2.5</sub> nonattainment areas, **a considerable amount if not a tremendous increase**.
- NH<sub>3</sub> from CCS may be **burdensome for PM<sub>2.5</sub> nonattainment regions** targeting 1-2 μg/m<sup>3</sup> reductions.
- If not properly controlled, amine scrubbing CCS may **seriously compromise the CCS social benefits** from CO<sub>2</sub> reductions.
- Since 60% of the social health costs occur during the winter, seasonal regulation could be considered.

## 7. Social Health Risks and Economic Valuations

- Software: BenMAP 4.0 developed by US EPA.
- Health Endpoint: Premature death from PM<sub>2.5</sub>.
- Value of a Statistical Life (VSL): \$8 millions (in 2010\$).

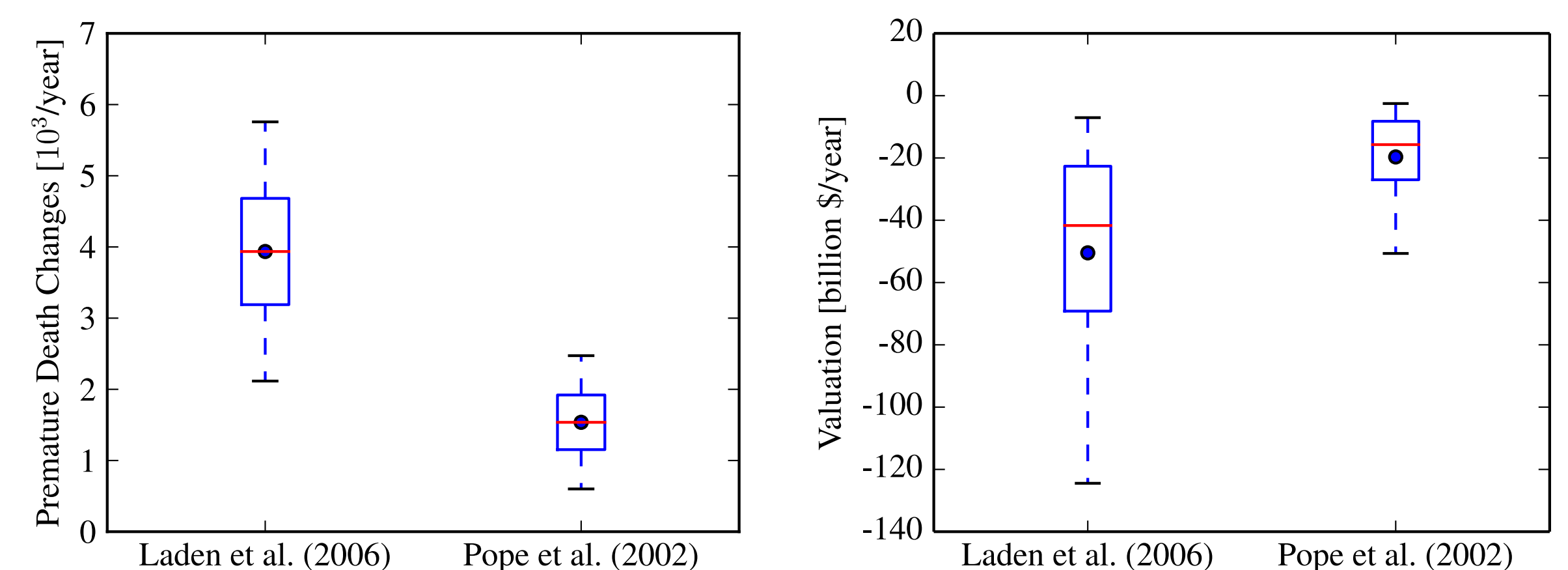


Figure 6: Social health risks of PM<sub>2.5</sub> increase from CCS NH<sub>3</sub> in 2050. Only uncertainties surrounding the CR functions and VSL are represented.

Table 3: Social health costs of CCS NH<sub>3</sub> and CO<sub>2</sub>.

CR function	2010\$/t NH <sub>3</sub>	2010\$/t CO <sub>2</sub>
Laden et al. (2006)	120,000	28
Pope et al. (2002)	46,000	11

## References

- Rochelle, G. T. *Science* **2009**, 325, 1652-1654.
- Rao, A. B.; Rubin, E. S.; Berkenpas, M. B. Department of Engineering and Public Policy, Carnegie Mellon University: Pittsburgh, PA 15213, **2004**.
- Ansari, A. S.; Pandis, S. N. *Environmental Science & Technology* **1998**, 32, 2706-2714.

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